

Gianluca Giorgi* and Laura Salvini

Centro Interdipartimentale di Analisi e Determinazioni Strutturali, Università di Siena,
via Aldo Moro I-53100, Siena, Italy

Fabio Ponticelli

Istituto di Chimica Organica, Università di Siena, via Aldo Moro I-53100, Siena, Italy

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Hydrolysis of 2-methylthiazolo[5,4-*b*]pyridine resulted in ring opening of the thiazole and formation of 3-acetamido-2(1*H*)-pyridinethione whose X-ray crystal structure has been determined.

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Introduction.

Thiazole derivatives have received and continue to receive considerable attention [1-4]. One interesting aspect of their chemistry concerns their photochemical reactions [5]. Another area of interest is the use of isothiazole dioxides as chiral auxiliaries in the stereoselective synthesis as well as their use as sweetening agents [6,7]. In addition, furanosylisothiazoles are involved in the chemistry of nucleic acids [8]. Isothiazole derivatives exhibit biological activities against wide classes of fungi and bacteria [9]. They can also act as inhibitors of blood platelet aggregations [10] and can have other important pharmacological properties, such as antiviral and anti HIV-1 activity [11].

For some years we have been developing a research project aimed at the synthesis and structural characterization of heterocyclic isomers formed by a pyridine ring fused with a five-membered ring [12,13]. In this project the eight members of 3-methyl-1,2- and 2-methyl-1,3-oxazolopyridines have been synthesized and characterized [14-16].

Recently our interest has been focused on thiazoles. We have synthesized and characterized all the members of the two classes of 3-methyl-1,2- and 2-methyl-1,3-thiazolopyridine isomers [17]. We report here the crystal and molecular structure of 3-acetamido-2(1*H*)-pyridinethione [18] obtained as hydrolysis product of 2-methyl-1,3-thiazolo[5,4-*b*]pyridine (Scheme 1).

Results and Discussion.

Among different reaction products possible, the addition of a water molecule to the 2-methylthiazolo[5,4-*b*]pyridine results in thiazole ring opening and the formation of 3-acetamido-2(1*H*)-pyridinethione. Crystal data are given in Table 1. Atomic coordinates and equivalent isotropic thermal parameters, as determined by X-ray crystallography, are reported in Table 2. The molecular structure is depicted in Figure 1, while geometrical parameters are given in Table 3.

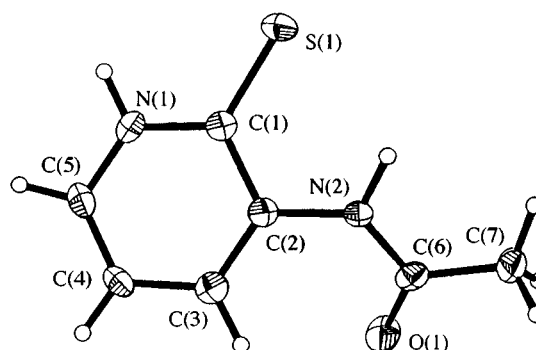


Figure 1. Molecular structure of 3-acetamido-2(1*H*)-pyridinethione. Thermal ellipsoids enclose 50% probability.

The C(1)=S(1) bond length is equal to 1.690(4) Å. This distance reflects the influence of substitution in position 3 as well as that on N(1). In different structure determina-

Scheme 1

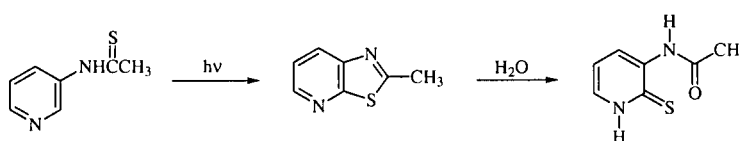


Table 1
Crystal Data for 3-Acetamido-2(1*H*)-pyridinethione

| | |
|--|---|
| Formula | C ₇ H ₈ N ₂ OS |
| M | 168.2 |
| Crystal size/(mm) | 0.3 x 0.2 x 0.2 |
| Crystal System | Orthorhombic |
| Space Group | Pna2 ₁ (n. 33) |
| <i>a</i> /Å | 7.000(1) |
| <i>b</i> /Å | 11.166(1) |
| <i>c</i> /Å | 9.770(1) |
| <i>U</i> /Å ³ | 763.6(1) |
| <i>Z</i> | 4 |
| F(000) | 352 |
| D _c /g cm ⁻³ | 1.46 |
| μ(Mo-Kα)/cm ⁻¹ | 0.361 |
| Radiation | graphite monochr. MoKα (λ = 0.71073 Å) |
| Scan mode | ω |
| Scan range/° | 2.8 ≤ θ ≤ 25.0 |
| Scan width/° | 0.90 |
| Scan speed/° min ⁻¹ | 3.0 |
| Temperature /°C | -93 |
| Unique reflections | 825 (R _{int} = 0.02) |
| Refinement Method | full-matrix anisotropic least squares on F ² for all reflections |
| No. parameters refined | 108 |
| R ₁ (F ² > 2σ(F ²)) | 0.036 |
| wR ₂ (F ² > 2σ(F ²)) | 0.063 |

Table 2
Atom Coordinates and Equivalent Isotropic Thermal Parameters (Å²)
for Non-hydrogen Atoms for 3-Acetamido-2(1*H*)-pyridinethione

| Atom | <i>x</i> / <i>a</i> | <i>y</i> / <i>b</i> | <i>z</i> / <i>c</i> | U _{eq} |
|------|---------------------|---------------------|---------------------|-----------------|
| S(1) | 0.07302(16) | -0.00573(8) | 0.51758(16) | 0.0300(3) |
| C(1) | 0.1068(6) | 0.1237(3) | 0.6040(4) | 0.0190(9) |
| N(1) | 0.1713(5) | 0.1221(3) | 0.7371(3) | 0.0222(9) |
| C(2) | 0.0728(6) | 0.2403(3) | 0.5489(4) | 0.0185(10) |
| C(3) | 0.1035(6) | 0.3410(4) | 0.6250(4) | 0.0234(10) |
| C(4) | 0.1673(6) | 0.3312(4) | 0.7625(4) | 0.0243(11) |
| C(5) | 0.2004(6) | 0.2205(3) | 0.8136(4) | 0.0246(10) |
| N(2) | 0.0042(5) | 0.2441(3) | 0.4130(3) | 0.0201(7) |
| C(6) | 0.0316(6) | 0.3344(4) | 0.3231(4) | 0.0216(10) |
| O(1) | 0.1267(5) | 0.4240(2) | 0.3496(3) | 0.0353(8) |
| C(7) | -0.0634(7) | 0.3187(4) | 0.1859(4) | 0.0307(12) |

tions of pyridine-2(1*H*)-thione this value ranges from 1.672 [19] to 1.700 Å [20], the last structure has been obtained at 116 K. In structures of substituted pyridine-2(1*H*)-thione derivatives, the lowest value determined for the C=S bond length is 1.660 Å found in 5-ethoxycarbonyl-6-methyl-4-(4-fluorophenyl)-3-cyano-2(1*H*)-pyridinethione [21], while the highest value is 1.719(6) Å found in the fused system *N*-(2-[5-methyl-1,3,4-thiadiazolo[3,2-*a*]pyridinio]) acetamidate [22].

The C(1)-N(1) and N(1)-C(5) bond lengths are 1.377(5) and 1.344(5) Å, respectively, in agreement with the mean values of 1.378 (σ_{mean} = 0.004) and 1.355 Å (σ_{mean} = 0.002) obtained for 11 X-ray structures of 3-substituted pyridine-2(1*H*)-thiones found from a search at the Cambridge Structural Database v. 5.16. The other C-C

Table 3
Bond Lengths (Å) and Angles (°) for
3-Acetamido-2(1*H*)-pyridinethione

| | |
|----------------|----------|
| S(1)-C(1) | 1.690(4) |
| C(1)-N(1) | 1.377(5) |
| C(1)-C(2) | 1.429(5) |
| C(2)-C(3) | 1.365(6) |
| C(3)-C(4) | 1.420(6) |
| C(4)-C(5) | 1.353(6) |
| C(5)-N(1) | 1.344(5) |
| C(2)-N(2) | 1.412(5) |
| N(2)-C(6) | 1.351(5) |
| C(6)-O(1) | 1.229(5) |
| C(6)-C(7) | 1.507(6) |
| S(1)-C(1)-N(1) | 120.4(3) |
| S(1)-C(1)-C(2) | 124.6(3) |
| N(1)-C(1)-C(2) | 115.0(3) |
| C(1)-C(2)-C(3) | 121.3(3) |
| C(2)-C(3)-C(4) | 120.1(4) |
| C(3)-C(4)-C(5) | 118.2(4) |
| C(4)-C(5)-N(1) | 121.0(3) |
| C(5)-N(1)-C(1) | 124.3(3) |
| C(1)-C(2)-N(2) | 116.0(3) |
| C(3)-C(2)-N(2) | 122.7(3) |
| C(2)-N(2)-C(6) | 125.8(3) |
| N(2)-C(6)-O(1) | 123.2(4) |
| N(2)-C(6)-C(7) | 115.4(3) |
| O(1)-C(6)-C(7) | 121.4(4) |

bond lengths in the pyridine ring are similar to those found in analogous structures. The C(2)-N(2) bond length is equal to 1.412(5) Å, while in the acetamido moiety the bond distances N(2)-C(6), C(6)-O(1) and C(6)-C(7) are equal to 1.351(5), 1.229(5), and 1.507(6) Å, respectively.

The S(1)-C(1)-N(1) and N(1)-C(1)-C(2) bond angles are equal to 120.4(3) and 115.0(3)°, respectively, in agreement with analogous structures. The exocyclic bond angle C(1)-C(2)-N(2) is equal to 116.0(3)°.

The pyridine ring is planar with atom deviations less than 0.008 Å from its least squares plane. The sulfur atom is almost coplanar with the heterocyclic system with a deviation of -0.015(1) Å from its least squares plane. The substituent in position 3 is also planar. The least squares plane defined by N(2), C(6), O(1) and C(7) forms a dihedral angle with the pyridine equal to 29.8(1)°. The torsion angle C(1)-C(2)-N(2)-C(6) is equal to -152.2(4)°.

Intramolecular interactions occur between the sulfur and H(N) atoms with S...H(N(1)) and S...H(N(2)) distances equal to 2.688(2) and 2.508(1) Å, respectively. Stacking interactions occur between molecules related by *x*+1/2, -*y*+1/2, *z* whose pyridine rings are almost parallel with intermolecular distances in the range 3.2-3.5 Å (Figure 2). The pyridine rings of adjacent molecules in the crystal form a dihedral angle equal to 38°. Intermolecular hydrogen bonds are also present in the crystal structure involving O(*x*, *y*, *z*)...H(N(1))(-*x*+1/2, *y*+1/2, *z*-1/2) and O(*x*, *y*, *z*)...H(N(2))(*x*+1/2, -*y*+1/2, *z*) with distances equal to 2.037(3) and 2.525(3) Å, respectively (Figure 2).

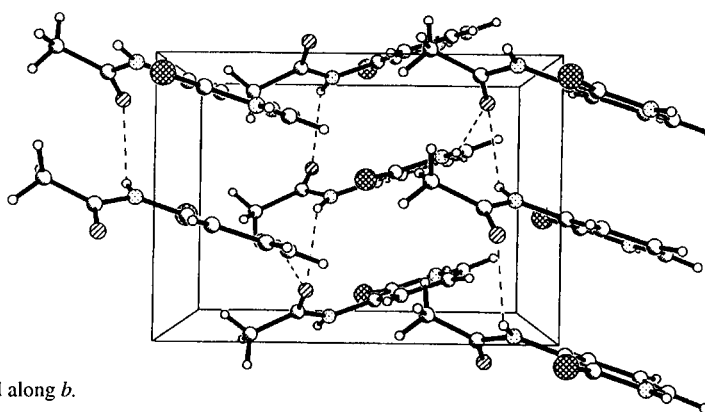


Figure 2. Crystal packing viewed along *b*.

EXPERIMENTAL

2-Methylthiazolo[5,4-*b*]pyridine was synthesized by photocyclization of 3-thioacetamidopyridine in hexane [23]. The addition of water yielded 3-acetamido-2(1*H*)-pyridinethione; m.p. 196-197° C.

Anal. Calcd. for C₇H₈N₂OS: C, 49.98; H, 4.79; N, 16.65; Found: C, 49.60; H, 4.42; N, 15.80.

Crystallographic data were collected on a Siemens P4 four-circle diffractometer with graphite monochromated Mo-K α radiation ($\lambda = 0.71069 \text{ \AA}$) in the range $-3 \leq h \leq 8$, $-13 \leq k \leq 1$, $-11 \leq l \leq 11$, ω scan mode for $2.8 \leq \theta \leq 25.0^\circ$ scan range, scan width 0.90° , constant scan speed $3.0^\circ \text{ min}^{-1}$. The data collection was performed at -93° C by using a Siemens LT-2A low temperature device. The temperature measured by a thermocouple was constant within $\pm 1^\circ \text{ C}$. Three standard reflections measured every 97 reflections showed no variations. 1042 total reflections ($R_{\text{int}} = 0.02$) were collected. Absorption correction was not applied.

The structure was solved by direct methods implemented in SHELX-97 [24]. The refinement was carried out by full-matrix anisotropic least-squares on F^2 for all reflections for all non-H atoms by using SHELX-97 [24]. The hydrogen atoms were localized in the Fourier map and included in the structure-factor calculations.

Final refinement of 108 parameters gave $R_1 = 0.036$ and $wR_2 = 0.063$ for $F^2 > 2\sigma(F^2)$. Weighting scheme was $w = 1/[\sigma^2(F_o)^2 + 0.0288P^2 + 0.1243P]$, where $P = (F_o^2 + 2F_c^2)/3$. Minimum and maximum height in last $\Delta\rho$ map were -0.20 and 0.22 e\AA^{-3} , respectively. Atomic scattering factors including f' and f'' were taken from reference 24. Geometrical calculations and molecular graphics were performed by using PARST [25] and SHELXTL [26] packages, respectively.

REFERENCES AND NOTES

- [1] Comprehensive Heterocyclic Chemistry II, A. R. Katritzky and C. W. Rees, eds, Pergamon Press, New York, 1996, Vol 3, Chap. 5.
- [2] K. R. H. Wooldridge, *Advances in Heterocyclic Chemistry*, Academic Press, New York, 1972, Vol 14, p 2.
- [3] M. Davis, *Advances in Heterocyclic Chemistry*, Academic Press, New York, 1972, Vol 14, p 43.
- [4] G. Giorgi, *Targets in Heterocyclic Systems*, O. A. Attanasi and D. Spinelli, eds, Italian Society of Chemistry, Rome, 1999, Vol 2, p 471.
- [5] J. W. Pavlik, *Organic Photochemistry (Series: Molecular and Supramolecular Photochemistry)*, V. Ramamurthy and K. S. Schanze, eds, Marcel Dekker, New York, 1997, Vol 1, p 57.
- [6] B. Schulze and K. Illgen, *J. Prakt. Chem.*, **339**, 1 (1997).
- [7] B. C. Chen, C. K. Murphy, A. Kumar, R. T. Reddy, C. Clark, P. Zhou, B. M. Lewis, D. Gala, I. Mergelsberg, D. Scherer, J. Buckley, D. Dibenedetto and F. A. Davis, *Organic Synthesis*, R. K. Boeckman, ed, J. Wiley & Sons Inc., New York, 1996, Vol 73, p 159.
- [8] D. K. Buffel, B. P. Simons, J. A. Deceuninck and G. J. Hoornaert, in *Nucleic Acid Chemistry*, L. B. Townsend and R. S. Tipson, eds, J. Wiley & Sons Inc., New York, 1991, pt 4, p 155.
- [9] A. H. Albert and D. E. O'Brien, *J. Heterocyclic Chem.*, **17**, 385 (1980).
- [10] K. H. Baggaley, German Patent 1976, 2,718,707; *Chem. Abs.*, **88**, 50843q (1978).
- [11] S. T. Ingate, J. L. Marco, M. Witvrouw, C. Pannecouque and E. De Clercq, *Tetrahedron*, **53**, 17795 (1997).
- [12] S. Chimichi, R. Nesi, F. Ponticelli and P. Tedeschi, *J. Chem. Soc. Perkin Trans. 1*, 1477 (1990).
- [13] S. Chimichi, D. Giomi, P. Tedeschi and F. Ponticelli, *Synth. Commun.*, **23**, 73 (1999).
- [14] S. Chimichi, R. Nesi, F. Ponticelli and P. Tedeschi *Magn. Reson. Chem.*, **23**, 86 (1985).
- [15] F. Ponticelli, D. Giomi, S. Papaleo and P. Tedeschi, *Org. Mass Spectrom.*, **28**, 451 (1993).
- [16] G. Giorgi, F. Ponticelli, G. Czira and K. Vékey, *J. Am. Soc. Mass Spectrom.*, **6**, 962 (1995).
- [17] G. Giorgi, L. Salvini, F. Ponticelli and P. Tedeschi, *J. Heterocyclic Chem.*, **33**, 1895 (1996) and references therein.
- [18] S. Hagen, G. A. Ulsaker and K. Undheim, *Acta Chem. Scand. B*, **28**, 523 (1974).
- [19] B. R. Penfold, *Acta Crystallogr.*, **6**, 707 (1953).
- [20] J. G. Reynolds, S. C. Sendlinger, A. M. Murray, J. C. Huffman and G. Christou, *Inorg. Chem.*, **34**, 5745 (1995).
- [21] Yu. A. Sharanin, A. M. Shestopalov, L. A. Rodinovskaya, V. N. Nesterov, V. E. Shklover, Yu. T. Struchkov, V. K. Promonenkov and V. P. Litvinov, *Zh. Org. Khim.*, **22**, 2600 (1986).
- [22] A. Kakehi, S. Ito and Y. Hashimoto, *Bull. Chem. Soc. Jpn.*, **69**, 1769 (1996).
- [23] A. Couture and P. Grandelaudon, *Heterocycles*, **22**, 1383 (1984).
- [24] G. Sheldrick, SHELX-97, Rel. 97-2, Program for X-ray data diffraction, Gottingen University, 1997.
- [25] M. Nardelli, PARST, Rel. February 1998, *J. Appl. Cryst.*, **28**, 659 (1995).
- [26] SHELXTL™, Ver. 5, Siemens Industrial Automation Inc., Madison, WI, 1994.